Fluorescence Emission Excitation Matrices (EEMs): Identifying Signatures for Constituents of Concern

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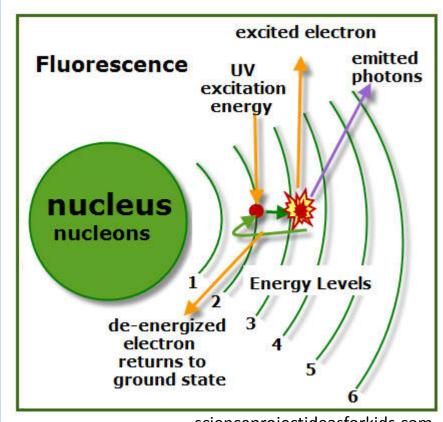
Outline

- What is spectrofluorescence and why use it?
 - Desire to find reliable, inexpensive methods to measure constituents
 - Brief Introduction to Spectrofluorescence
 - Application to measuring drinking water
 Constituents of Concern
- Spectrofluorometer Special Study
 - Goals, Study Design
 - Example Data
 - Progress & Next Steps

Spectrofluorescence Intro

 Fluorescence: When an electron is excited to a higher energy level (electron orbit) by absorption of light energy, and then releases energy as light

as it drops to a lower energy level.



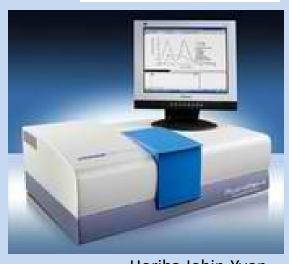
scienceprojectideasforkids.com

Intro, continued: Instruments

- A fluorometer: one pair of light wavelengths to measure, e.g., chlorophyll in algal cells.
 - Turner 10AU,
 - in-situ FDOM probe

- A spectrofluorometer: Performs multiple measurements across bands of light wavelengths.
 - Horiba/Jobin-Yvon Fluoromax-4 purchased by QAQC Program.

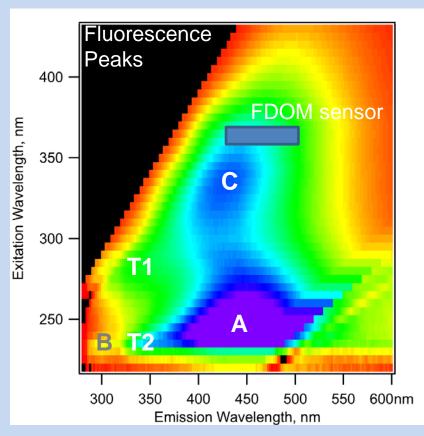




Horiba Jobin-Yvon

Data: Excitation-Emission Matrix

Each excitation-emission matrix (EEM) consists of hundreds of measurement combinations of a single water sample, with excitation wavelength is on one axis, emission wavelength is the second, and fluorescence intensity forms a third axis.



Spectrofluorescence Special Study

- To investigate the usefulness of spectrofluorescence as a way of quickly and easily quantify constituents of concern (CoCs) in source waters.
 - Organic carbon: Demonstrated
 - Nitrosamines: Possible?
 - Hua et al., 2007, Fluorescence fingerprints to monitor total trihalomethanes and N-nitrosodimethylamine formation potentials in water. Environ Chem. Lett. 5:73–77.
- Source water "fingerprinting"
 - Find distinctive fluorescence features in sources

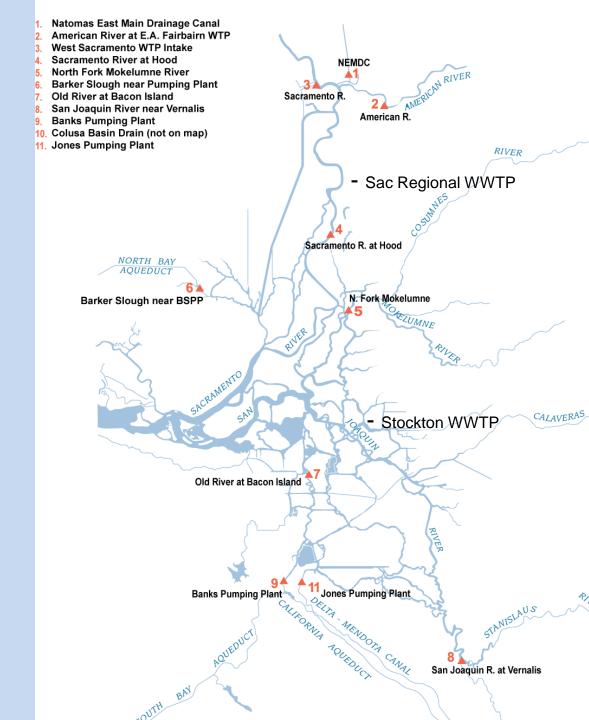
Study Design

- Two-year study, approved in the Work Plan
 - 11 stations sampled monthly (piggy-backed on MWQI routine sampling program), plus several other sites sampled at other frequencies.
 - Sample analysis for DOC, THMFP, HAAFP, Nitrogen chemical species, Spectrofluorescence EEM, Nitrosamine formation (~quarterly sampling).
 - Numerical analysis to identify features in the EEMs that correlate highly with Constituents of Concern.

Monthly Sampling Stations

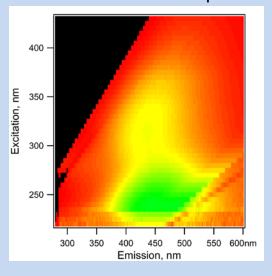
Represents:

- Major tributaries
- Seasonal variation
- Spatial variation
- Diff. Source waters
 - AMR "Pristine"
 - Colusa Ag runoff
 - NEMDC Urban
 - Waste Water

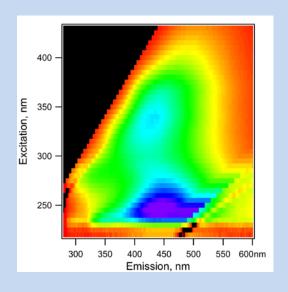


Example EEM Data: Spatial Variation

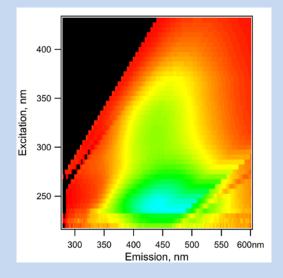
San Joaquin R, Downstream of Stockton WWTP 12 Sept 2011



Stockton WWTP 05 Oct 2011



Natomas East
Main Drain
(NEMDC)
07 Mar 2011



Example EEM Data: Spatial Variation

1.705 E+06

1.281E+06

8.575 E+05

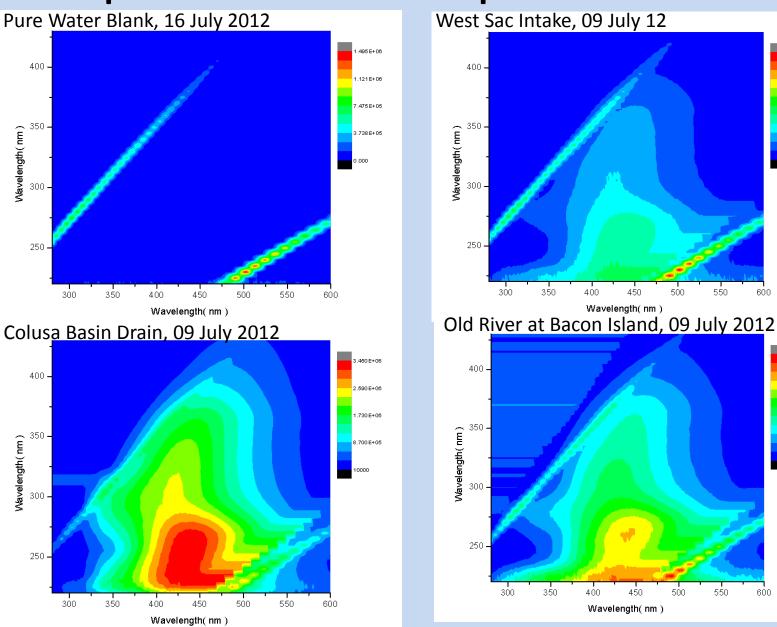
4.338 E+05

635 E±06

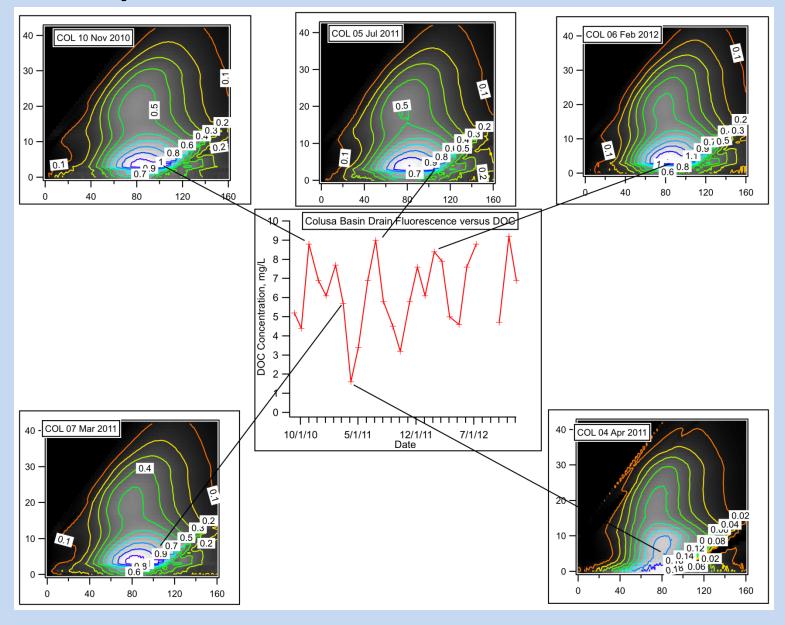
1.100 E+06

5.650 E+05

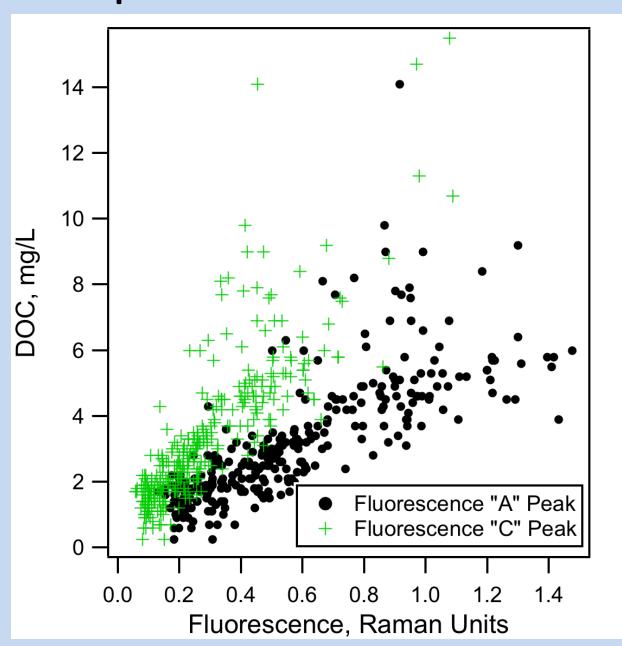
.000 E+04

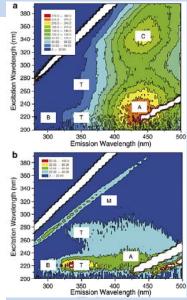


Example EEM Data: Time variation

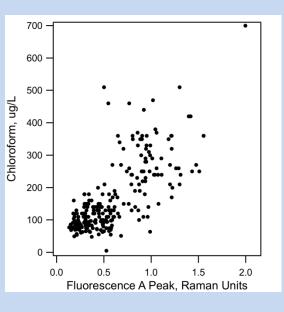


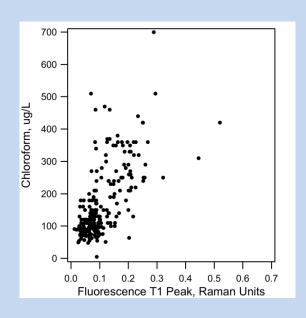
Example EEM Data: Correlation

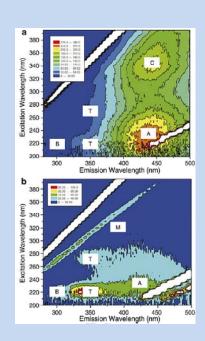


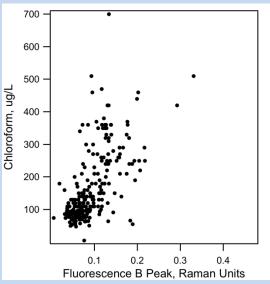


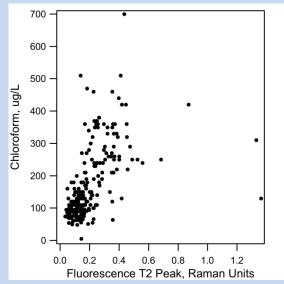
Chloroform and EEM features





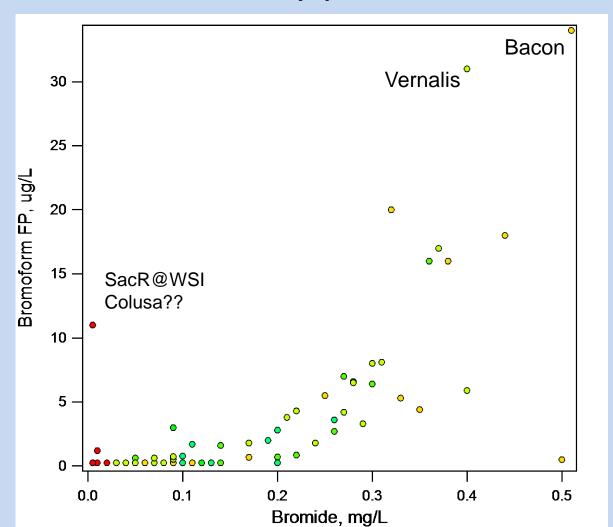


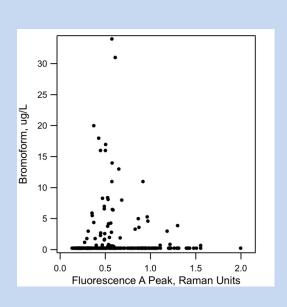




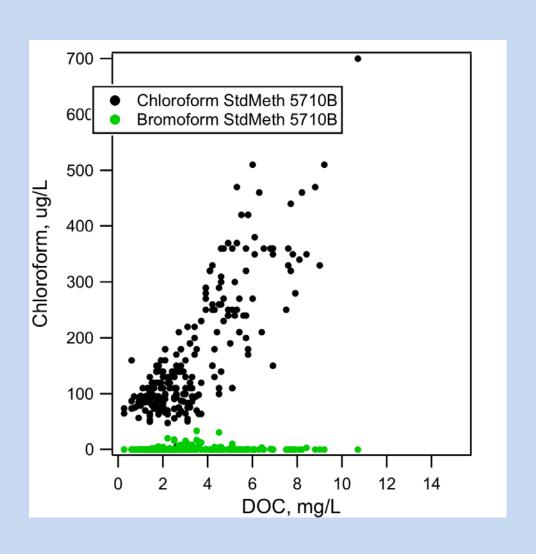
Bromoform and EEM features

- Mostly a function of source waters
- EEMs not very predictive of Bromoform FP

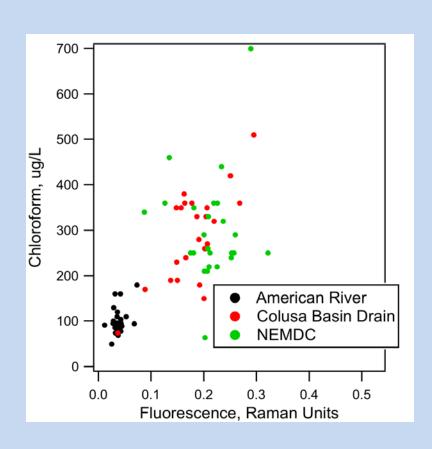


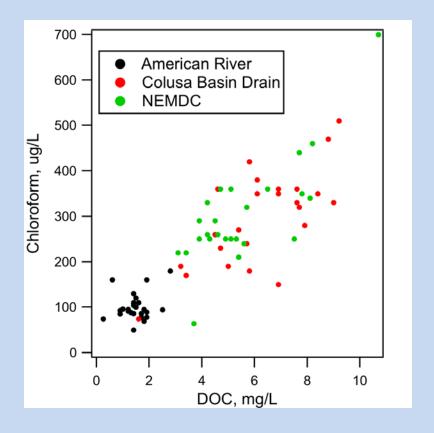


Halogens vs DOC

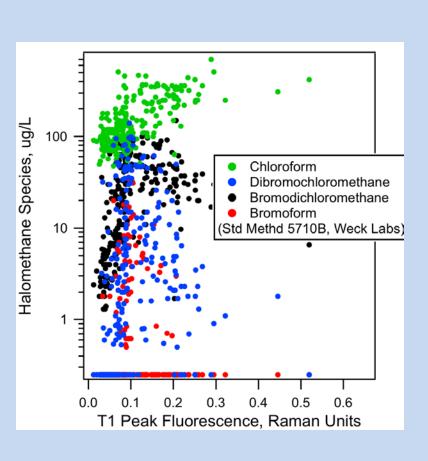


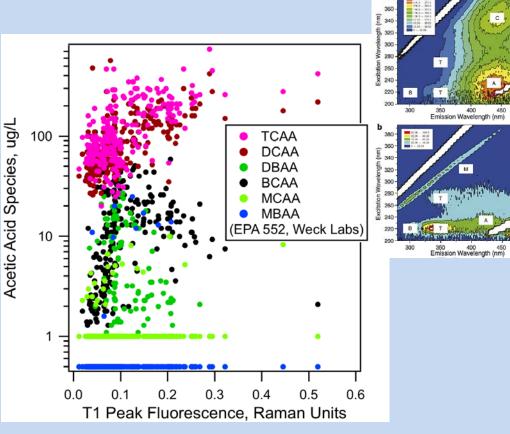
Chloroform variation



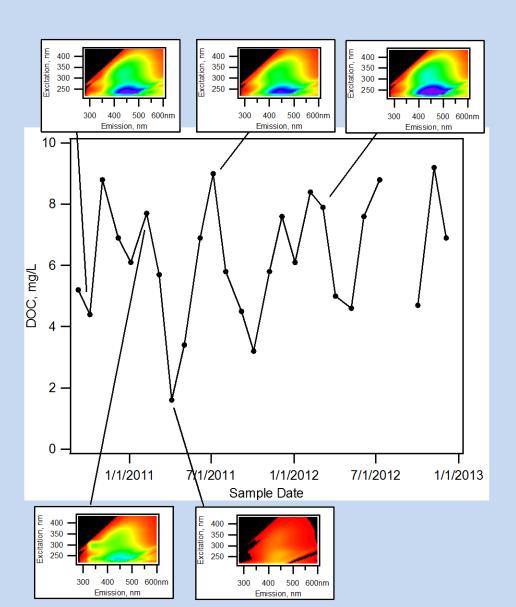


THMs & HAAs vs T1 Fluorescence

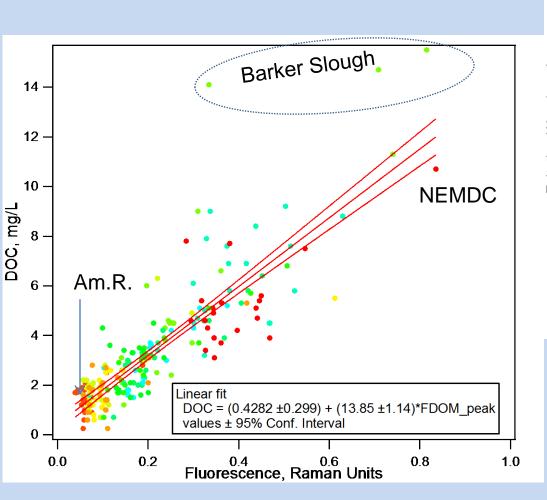


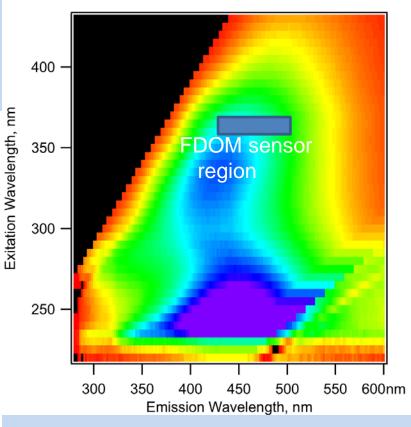


DOC



Likely FDOM sensor response





Progress

- Successfully analyzing monthly water samples from 11 monthly + 8 other sites for Spectrofluorescence EEM, DOC, THMFP, HAAFP, Nitrogen chemical species likely to be nitrosamine precursors.
 - Nitrosamine correlation pending.
- Products: Final study report, feasibility for future monitoring, peer-reviewed article

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- Craig Nelson, UC Santa Barbara: software adaptation
- Stuart Krasner, MWD: Nitrosamine collaboration
- Bryte Laboratory, Weck Laboratory: CoC lab analysis

Questions & Comments?

References

Selected References

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